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Analytical chemistry studies and uses instruments and methods used to separate, identify, and quantify matter. In practice, separation, identification or quantification may constitute the entire analysis or be combined with another method.

Analytical chemistry - Wikipedia

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Gravimetric analysis describes a set of methods used in analytical chemistry for the quantitative determination of an analyte (the ion being analyzed) based on its mass. The principle behind this type of analysis is that once an ion's mass has been determined as a unique compound, that known measurement can then be used to determine the same analyte's mass in a mixture, as long as the relative ...

Gravimetric analysis - Wikipedia

Chapter 5 36 Complexometric Titrations Complexometric titrations are based on the formation of a soluble complex upon the reaction of the species titrated with the titrant.

Complexometric Titrations - cffet.net

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Espectroscopia UV/visã•vel â€“ Wikipã•dia, a enciclopã•dia livre

Los instrumentos de control estã;n universalmente aceptados. Hoy en dã•a es inimaginable la existencia de una industria moderna sin instrumentos.

Instrumentación Industrial â€“ Antonio Creus â€“ 6ta Edición

El coeficiente de actividad γ_i de un componente i en una solución real se define como el cociente entre su actividad real a_i y su actividad ideal a_i^* en la misma temperatura y presión. En el caso de un electrolito fuerte, la actividad real se relaciona con la concentración de los iones a través de la constante de disociación K_a y el coeficiente de actividad de los iones γ_{\pm} . La actividad de un electrolito fuerte se expresa como $a_{\pm} = \gamma_{\pm} m_{\pm}$, donde m_{\pm} es la molalidad de los iones. El coeficiente de actividad de los iones γ_{\pm} se relaciona con el coeficiente de actividad del electrolito γ_{\pm} a través de la relación $\gamma_{\pm} = \gamma_{\pm}^{\nu_+ \nu_-}$, donde ν_+ y ν_- son los coeficientes estequiométricos de los iones positivos y negativos, respectivamente. El coeficiente de actividad de los iones γ_{\pm} se relaciona con el coeficiente de actividad del electrolito γ_{\pm} a través de la relación $\gamma_{\pm} = \gamma_{\pm}^{\nu_+ \nu_-}$.

El coeficiente de actividad γ_i de un componente i en una solución real - El coeficiente de actividad γ_{\pm} de un electrolito fuerte

La difusión de la materia, o difusión química, designa la tendencia natural de un sistema a hacer uniforme el potencial químico de cada una de las especies químicas que lo componen. Es un fenómeno de transporte irreversible que se traduce por la migración de especies químicas en un medio. Bajo el efecto de la agitación térmica, se observa un desplazamiento de los constituyentes ligados...

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